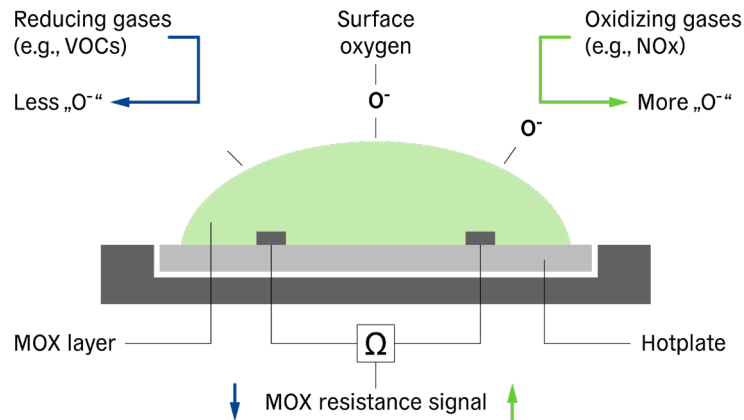


What is a metal oxide (MOX) sensor?

Learn about the possibilities and limitations of MOX sensors and the strength of Sensirion's SGP4x products



A MOX sensor reacts to most changes in gas composition

In a nutshell, a MOX sensor is a heated surface of a metal oxide that changes its electrical resistance depending on the oxygen content on its surface. Oxidizing gases like NO_x (providing more oxygen than ambient air) increase the resistance, whereas reducing gases like VOCs (consuming oxygen by being combusted on the metal oxide surface) reduce the resistance. Humidity also impacts the MOX sensor signal, as water vapor usually behaves as a reducing gas. This can be compensated for by using a humidity sensor such as Sensirion's SHTxx. Sensirion's SGP4x sensors feature on-chip humidity compensation.

MOX sensors are excellent devices to detect relatively short-term changes in gas compositions and better understand the activities associated with gas events in indoor environments.

MOX sensors are semiquantitative and broadband-sensitive

By calibrating the MOX resistance to a specific target gas, under laboratory conditions the absolute concentration of the target gas in air

can be measured. However, under field conditions, two aspects hamper MOX sensors' ability to provide absolute concentration outputs.

Firstly, a well-defined baseline is required. A baseline is a reference point for a well-known concentration of the target gas. Since the signal of a MOX sensor usually drifts over time, this must be compensated for in the field. However, under field conditions, it cannot be ensured that the sensor is exposed to the exact same concentration it has been calibrated for, leading to an erroneous concentration output.

Secondly, a MOX sensor is a broadband-sensitive device, meaning that it reacts to multiple gases and cannot distinguish between them. Thus, calibrating such a sensor to a specific gas or gas mixture – as it is usually the case with digital MOX sensors – does not make these devices selective to this particular gas (mixture). For instance, a calibrated sensor could show 1000 ppb as the output, whether it is related to 500 ppb of ethanol or 2000 ppb of toluene. However, through material design and operation mode engineering, it is possible to enhance the selectivity to specific gas groups – namely, reducing and oxidizing gases – to distinguish between VOC and NO_x events, as it was done with Sensirion's SGP41.

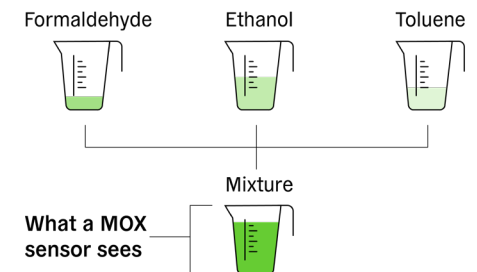
Nevertheless, because MOX sensors are sensitive and fast, they can reliably report short-term changes in the gas composition relative to a certain period in history (e.g., the past 24 hours).

Further reading

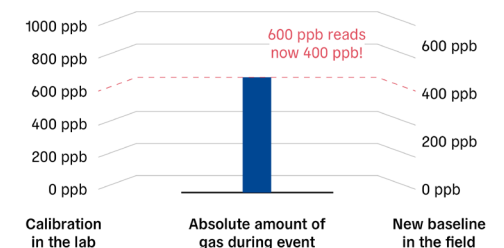
[What are reducing gases?](#)

[What are oxidizing gases?](#)

How the real gas composition looks like:



For MOX sensors, a change in baseline changes the output scale



Figures above demonstrate the limitation to provide a concentration output for a specific target gas by a MOX sensor due to broadband sensitivity (top) and varying baseline conditions in the field (bottom).